

Application Note

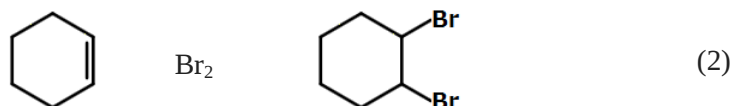
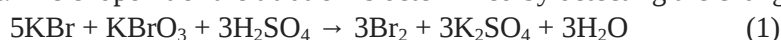
Bromine number of cyclohexene

Industry	:	Petroleum
Instrument	:	Automatic potentiometric titrator
Measurement method	:	Polarization titration at constant current
Standards	:	JIS K2605, GB/T 11135

1. Scope

Bromine number of cyclohexene was measured by polarization titration at constant current based on “JIS K2605 Petroleum distillates and commercial aliphatic olefins - Determination of bromine number - Electrometric method”.

Bromine number is the number of grams of bromine that is added to carbon-carbon unsaturated bond in a 100 g of sample and is expressed in “gBr₂/100g”. In measurement of bromine number, bromine is released by adding potassium bromide- potassium bromate (KBr-KBrO₃) solution into a sample solution containing strong acid, and then the released bromine is added to carbon-carbon double bond. The chemical formulas of bromine releasing reaction and addition reaction are given in formula (1) and (2). During titration, polarization potential which needs to apply constant current between twin platinum electrodes is measured and, an equivalent point is determined by change of the polarization potential. After the equivalent point, amount of bromine in the solution is increased sharply. Polarization potential becomes smaller when amount of bromine in the solution become larger, therefore a sharp change in potential is observed at the equivalent point. The endpoint of the titration is determined by detecting the change in potential.



In this measurement, toluene solution of cyclohexene was added to titration solvent whose temperature was kept in a range of 0-5°C, and then the solution was titrated with 0.25 mol/L KBr-KBrO₃ solution. An inflection point on the titration curve was regarded as the end point, and the bromine number of cyclohexene was calculated from the volume of KBr-KBrO₃ solution consumed to titrate sample to the end point.

2. Precautions

- 1) Handle the reagents in a well ventilated room or a draft chamber.
- 2) Keep a temperature of titration solvent in a range of 0-5°C during titration.
- 3) Discharge titrant between burette and titration nozzle once before a titration.

3. Post-measurement procedure

Wash electrode and titration nozzle with ethanol and pure water and then keep them soaked in pure water. If titration nozzle is left standing in the air, it will be clogged with precipitate of the component of titrant.

4. Apparatus

Main unit	:	Automatic potentiometric titrator (preamplifier : POT)
Electrode	:	Twin platinum electrodes Temperature compensation electrode

Titrant : 0.25 mol/L KBr-KBrO₃ solution
 Titration solvent : Mixture of glacial acetic acid, toluene, methanol and H₂SO₄(1+5)
 mixture ratio of them is 714 : 134 : 134 : 18 (volumetric ratio).

-Preparation of sample solution-

- 1) Add 10 mL of toluene into a 50 mL volumetric flask.
- 2) Add 0.6 - 1.0 g of cyclohexene into the flask and measure mass of it.
- 3) Add toluene to the mark of the flask and mix it.

-Setting of constant current value-

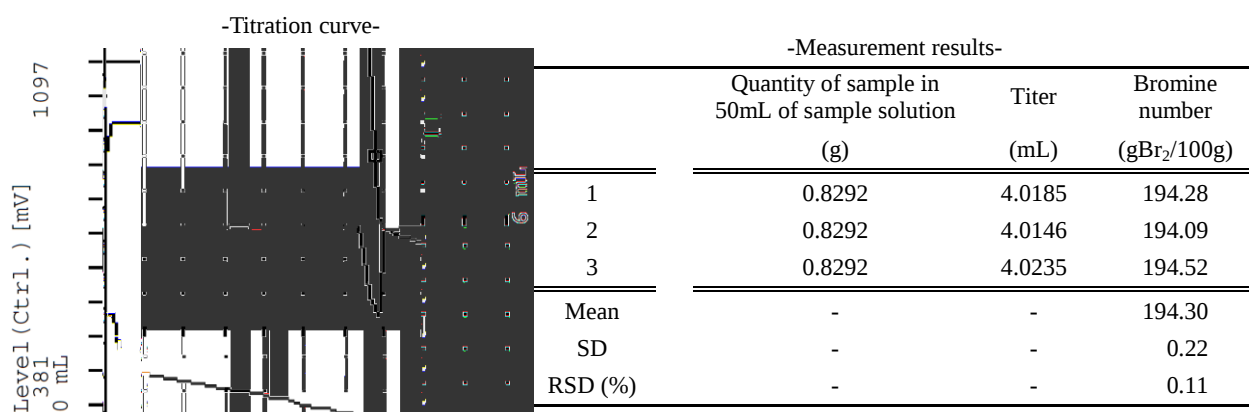
- 1) Press [Calibration] button.
- 2) Set Channel/Unit to "Ch3/Pol".
- 3) Press [Details] button and set as follows.
 Calibration Mode Current Polar Current 5.00 µA
- 4) Immerse twin platinum electrode in titration solvent and perform calibration.

-Blank test-

- 1) Add 110 mL of titration solvent into a 200 mL tall beaker.
- 2) Keep temperature of the titration solvent in a range of 0-5°C by ice cooling.
- 3) Add 5 mL of toluene.
- 4) Titrate with 0.25 mol/L KBr-KBrO₃ solution. at]

- 1) Add 110 mL of titration solvent into a 20 mL tall beaker.
- 2) Keep temperature of the titration solvent in a range of 0-5°C by ice cooling.
- 3) Add 5 mL of toluene.

-Results-



9. Summary

In this measurement, the results showed a good repeatability with 0.11% RSD (relative standard deviation), and the results of bromine number was within range specified in JIS K2605 (187-199 gBr₂/100g).

In some samples, verification of the measurement capability is required. In such case, please contact us.

10. References

- 1) JIS K2605-1996 Petroleum distillates and commercial aliphatic olefins - Determination of bromine number - Electrometric method